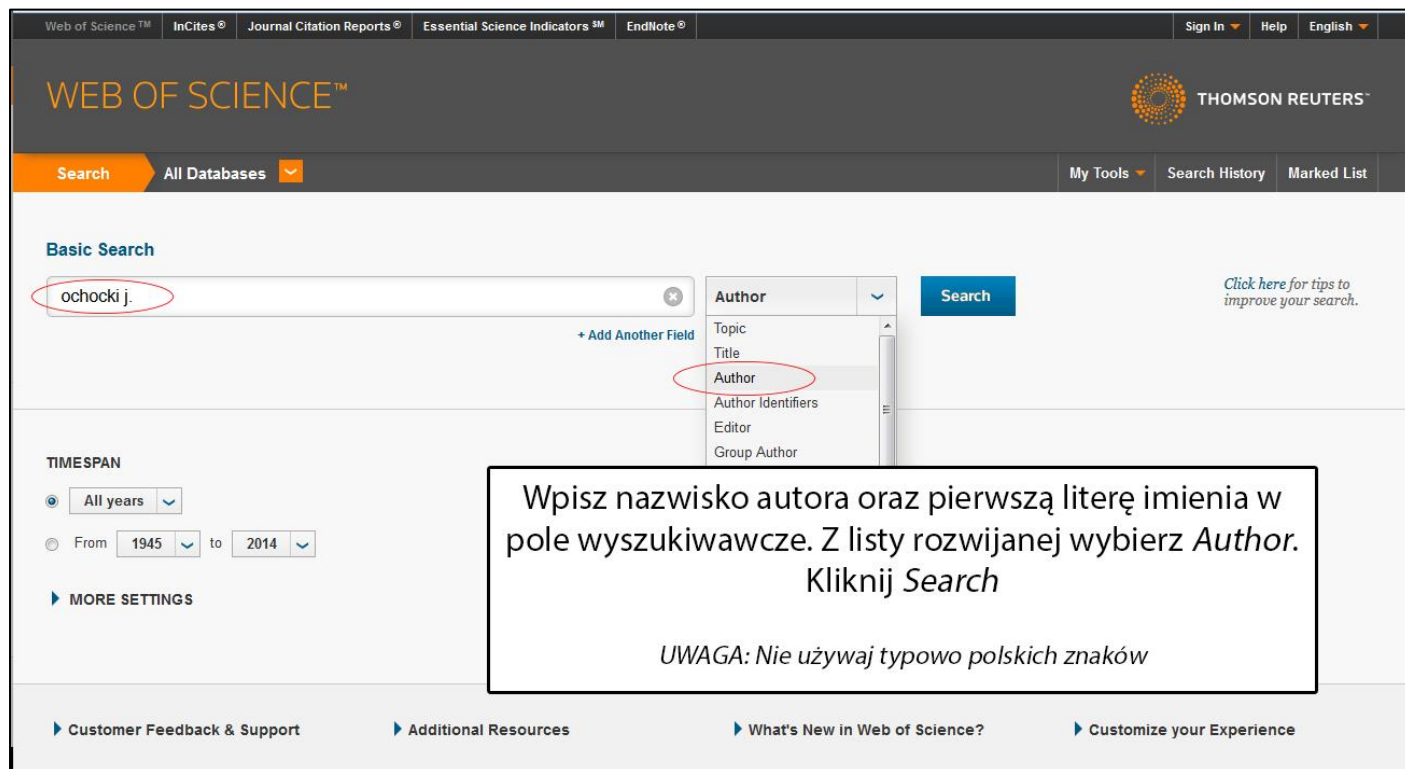


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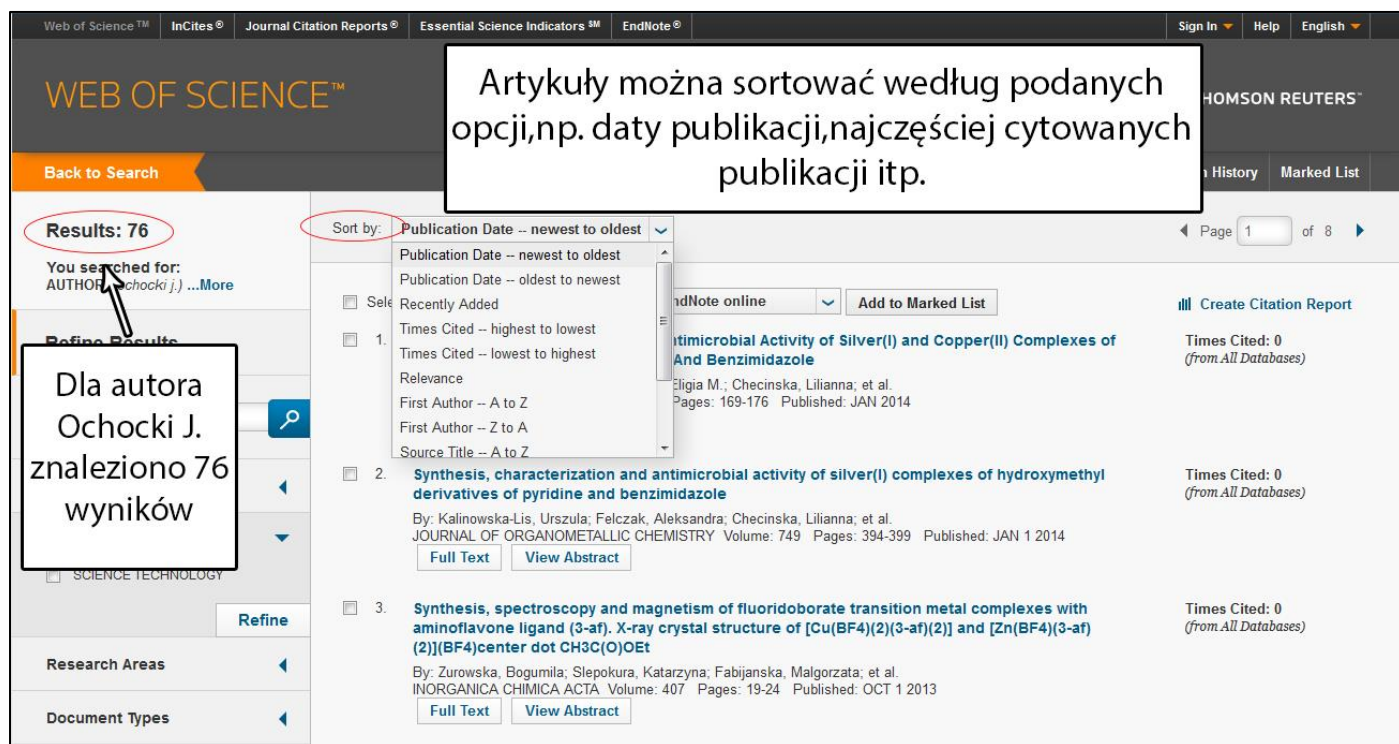
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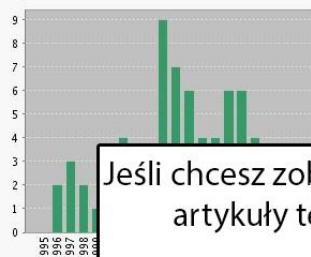
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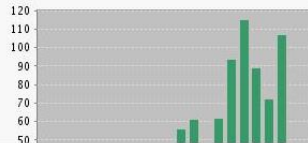
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The interaction of diethyl 2-pyridylmethylphosphonate Cu(2-pca)(2) [X = Cl- (1), CH3COO- (3)], and Cu(2-pca)(2-qmpe) reacts with CuX2 salts to similarly yield complexes Cu(2-qca)Cl (6), (2-qca = quinoline-2-carboxylate ion), 2-pmpe and 2-qmpe ligands under the used conditions. Cu(2-pca)(2) exists in two crystalline forms, a blue form (1) and a violet form (3). For 3 the single-crystal structure was determined. The copper atom is four-coordinated in a square-planar geometry. The stack between related (and hence parallel) pea moieties involves interatomic distances of 3.27 Angstrom. Cu(2-qca)(2) (.) H2O also exists in two forms, a green (4) and a blue-green (5). Both these complexes are five coordinated, involve the same CuN2O3 chromophore and are examples of the distortion isomers. Variable-temperature magnetic susceptibility measurements (1.9-300 K) have shown that the anti ferromagnetic coupling observed is much stronger in 6 than in 1, 3, 4 and 5. For 2 a ferromagnetic, exchange occurs. (C) 2003 Elsevier B.V. All rights reserved.

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